DIVISION

IN THE UNITED STATES PATENT AND TRADEMARK OFFICE

In re Patent Application of

ACTON et al

Serial No. 10/014,795

Filed: December 14, 2001

For: REDUCTION IN MINERAL SALT DEPOSITION

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2009 AUG 10 PH 2: 39

Atty. Ref.: 3948-4

TC/A.U.: 1754

Examiner: Kuhar, Anthony J.

August 5, 2004

Mail Stop 16 Commissioner for Patents P.O. Box 1450 Alexandria, VA 22313-1450

Sir:

REQUEST FOR REFUND

A refund is respectfully requested for the overpayment made with the filing of a Notice of Appeal, an Amendment and a Petition to Extend, filed on July 21, 2004 (copy enclosed). When the Notice of Appeal, Amendment and Petition to Extend were filed, applicant's attorney inadvertently listed "\$1,280.00 as the basic filing fees of this patent application, when in fact the correct amount is "\$640.00" for a small entity. Therefore it is requested that the overpayment of \$640.00 be refunded to applicant's undersigned attorney by crediting \$640.00 to our deposit account no. 14-1140 under Order No. 3948-

4. For this purpose a duplicate copy of this paper is enclosed.

Respectfully submitted,

NIXON & VANDERHYE P.C.

By:

Leonard C. Mitchard Reg. No. 29,009

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IN THE UNITED STATES PATENT AND TRADEMARK OFFICE Before the Board of Patent Appeals and Interferences

In re Patent Application of

Atty Dkt.: 3948-4 C#

2014 AUG 10 PM 2: 39

ACTON et al

Serial No. 10/014,795

Filed: December 14, 2001

Title:

REDUCTION IN MINERAL SALT DEPOSITION

Examiner: Kuhar, Anthony J.

TC/A.U.: 1754

Date: July 21, 2004

Commissioner for Patents P.O. Box 1450

Alexandria, VA 22313-1450

Sir:

⊠ Correspondence Address Indication Form Attached.

X	NOTICE O	F APPEAL ereby appeals to the Board of Pa	tent Anne	als and Interfe	rences from the l	ast decision	•	
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Any future submission requiring an extension of time is hereby stated to include a petition for such time extension. The Commissioner is hereby authorized to charge any deficiency, or credit any overpayment, in the fee(s) filed, or asserted to be filed, or which should have been filed herewith (or with any paper hereafter filed in this application by this firm) to our Account No. 14-1140. A duplicate copy of this sheet is attached.

1100 North Glebe Road, 8th Floor Arlington, Virginia 22201-4714 Telephone: (703) 816-4000 Facsimile: (703) 816-4100 LCM:lfm

NIXON & VANDERHYE P.C.

By Atty: Leonard C. Mitchard, Reg. No. 29,009

Signature:

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IN THE UNITED STATES PATENT AND TRADEMARK OFFICE

In re Patent Application of	Atty Dkt. 3948-4 DIVIS	אסול
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Serial No. 10/014,795	Examiner: Kuhar, Anthony J.	:
Filed: December 14, 2001	Date: July 21, 2004 Z004 AUS 10	PM 2: 39
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Commissioner for Patents P.O. Box 1450 Alexandria, VA 22313-1450 Sir:		· :
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Fees are attached as calculated below: Total effective claims after amendment previously paid for 24 (at least 20) =	minus highest number 0 x \$ 18.00	\$ 0.00
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Terminal disclaimer enclosed, add \$ 110.00		\$ 0.00
First/second submission after Final Rejection Please enter the previously unentered Submission attached	n pursuant to 37 CFR 1.129(a) (\$770.00) , filed	\$ 0.00
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Rule 56 Information Disclosure Statement Filing	Fee (\$180.00)	•
Assignment Recording Fee (\$40.00)		\$ 0.00
Other: Notice of Appeal		330.00
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1100 North Glebe Road, 8 th Floor Arlington, Virginia 22201-4714 Telephone: (703) 816-4000 Facsimile: (703) 816-4100 LCM:lfm	NIXON & VANDERHYE P.C. By Atty: Leonard C. Mitchard, Reg. No. 29,009 Signature:	

<u>AMENDMENTS TO THE CLAIMS:</u>

201 AUG 10 PM 2: 39

This listing of claims will replace all prior versions, and listings, of claims in the application:

1-7 (cancelled).

8 (currently amended). A method of reducing deposition of mineral salts from an aqueous supersaturated solution onto a solid surface in contact with the aqueous supersaturated solution which method consists essentially of:

- (a) forming a composition comprising a dispersion of either (i) seed crystals of the mineral salt in an aqueous solution of the mineral salt or (ii) seed crystals of a salt isomorphous with the mineral salt in an aqueous solution of the isomorphous salt wherein the dispersed seed crystals are of mean particle size of less than 2.5 microns and have been obtained via generation of cavitation in an aqueous supersaturated solution of the mineral salt or a salt isomorphous with the mineral salt;
- (b) distributing said composition into either (i) an aqueous supersaturated solution of the mineral salt or (ii) an aqueous precursor liquid of the aqueous supersaturated solution which aqueous precursor liquid is saturated with respect to the seeds, and in the case of (b)(ii) converting the aqueous precursor liquid into an aqueous supersaturated solution of the mineral salt; and by (i) cooling and/or reducing the pressure of the aqueous precursor liquid or (ii) by adding a complementary iron to the precursor liquid; and;
 - (c) contacting the aqueous supersaturated solution with the solid surface.

. 7

9 (original). A method as claimed in Claim 8 wherein the composition and 2: 39 obtained by subjecting an aqueous supersaturated solution of either (i) the mineral salt or (ii) a salt isomorphous with the mineral salt to sonic or ultrasonic vibration.

10 (original). A method as claimed in Claim 9 wherein the seed crystals have 0.025 – 0.5 times the diameter of crystals of the same mineral salt which crystallise out from an aqueous supersaturated solution thereof in the absence of sonic or ultrasonic vibration.

11 (previously presented). A method as claimed in claim 9 wherein the frequency of the ultrasonic vibration is between 16 and 40kHz.

12 (original). A method as claimed in Claim 11 wherein the energy density applied to the aqueous supersaturated solution by the ultrasonic vibration is in the range of from 1 to 100 J/cm³.

13 (previously presented). A method as claimed in claim 9 wherein the duration of the applied ultrasonic vibration is 0.05 to 360 seconds.

14 (previously presented). A method as claimed in claim 9 wherein the degree of supersaturation of the aqueous supersaturated solution which is subjected to the ultrasonic vibration is 50 to 400 times over the saturation level.

15 (previously presented). A method as claimed in claim 9 wherein the 2004 AUG 10 PM 2: 39 supersaturated solution which is subjected to the ultrasonic vibration is obtained by passing 2 or more aqueous solutions of the separate components of the mineral salt or of the separate components of the isomorphous salt to a locus of mixing, at which locus the aqueous supersaturated solution is formed and the ultrasonic vibration is applied.

16 (previously presented). A method as claimed in claim 8 wherein the percentage weight of seed crystals from the dispersion to the total weight of seed crystals and depositable mineral salts is in the range 10 to 50% w/w.

17 (currently amended). A method as claimed in claim 8 wherein distribution of the composition into the aqueous supersaturated solution or aqueous precursor liquid is performed 2 to 4 times with the distribution being in series or parallel or a combination of both.

18-24 (cancelled).

25 (previously presented). A method as claimed in Claim 8 wherein the mineral salt is selected from the group consisting of alkaline earth metal carbonates and alkaline earth metal sulphates.

26 (previously presented). A method as claimed in Claim 8 wherein the seed crystals are of Mean particle size of 0.5 to 2 microns.

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27 (previously presented). A method as claimed in Claim 8 wherein the seed crystals are present in the dispersion in an amount of from 1 to 60% by weight based on the total weight of dispersion.

28 (previously presented). A method as claimed in Claim 8 wherein the seed crystals are barium sulphate crystals and have 3 dimensional distances of length, breadth and thickness, normal to one another, in a ratio of 0.4-1.5:1:0.4-1.5.

29 (previously presented). A method as claimed in Claim 8 wherein the seed crystals have one or more voids therein which voids occupy 5 to 40% of the volume enclosed by the envelope of the outer surface of the seed crystals.

30 (previously presented). A method as claimed in Claim 8 wherein the seed crystals are rounded calcium carbonate crystals having a diameter in the range of 1 to 2.5 microns.

31 (cancelled).

32 (previously presented). A method as claimed in Claim 8 further comprising the step of monitoring the size of the crystals.

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REMARKS/ARGUMENTS

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Reconsideration of this application is requested. Claims 8-17, 25-30 and 32 are in the case.

I. THE OBVIOUSNESS REJECTIONS

Claims 8-16, 18, 25-29 and 32 stand rejected under 35 U.S.C. §103(a) as allegedly unpatentable over EP 0196,622 in view of Gatumel et al. That rejection is respectfully traversed.

As now claimed, the method of the invention is for reducing deposition of mineral salts from an aqueous supersaturated solution onto a solid surface in contact with the aqueous supersaturated solution. The method consists essentially of the steps of (a) forming a composition comprising a dispersion of either (i) seed crystals of the mineral salt in an aqueous solution of the mineral salt or (ii) seed crystals of a salt isomorphous with the mineral salt in an aqueous solution of the isomorphous salt wherein the dispersed seed crystals are of mean particle size of less than 2.5 microns and have been obtained via generation of cavitation in an aqueous supersaturated solution of the mineral salt or a salt isomorphous with the mineral salt; (b) distributing the composition into an aqueous precursor liquid of the aqueous supersaturated solution which aqueous precursor liquid into an aqueous supersaturated solution of the mineral salt by (i) cooling and/or reducing the pressure of the aqueous precursor liquid or (ii) by adding a complementary iron to the precursor liquid; and; (c) contacting the aqueous supersaturated solution with the solid surface.

A method of distributing a composition of seed crystals into an aqueous 2014 AUG 10 PH 2: 39 precursor liquid (which is not supersaturated) is not described or suggested by EP 916622, when take alone or in combination with Gatumel. EP 916622 discloses addition of a scale formation preventer (such as particles of calcium sulphate) to a composition which is already in a supersaturated condition. Thus, it can be seen from reference to Figure 1 and the Examples that the complementary ions, e.g. calcium and sulphate, are fed from tanks A and B into tank D where a supersaturated solution is formed (see page 5, lines 22-23). Calcium sulphate particles are then added into this supersaturated solution from tank C.

It has been found according to the present invention that with more highly reactive systems such as those comprising barium and sulphate ions, the method described in EP 916622 would not be very suitable and would still lead to a large amount of mineral salt deposition. This is because with highly reactive ions such as barium and sulphate, when these are mixed in the absence of seed crystals, spontaneous nucleation would occur very rapidly. Crystallisation (and therefore unwanted deposition of mineral salt) would occur within seconds, and this could not be prevented by the subsequent addition of seed crystals to the supersaturated solution (as occurs in EP 916622). This phenomenon is critical, particularly in the oil industry where high levels of supersaturation are encountered, and where barium and sulphate ions are routinely present. The method of EP 916622 would not be suitable or efficient enough under these conditions.

By contrast, in accordance with the amended claims presented herewith, in the present method the seed crystals are added to a precursor solution which is not

supersaturated with respect to either of the reactive ions. Only thereafter is all allowed to a mix downstream with a solution of complementary ions which will then lead to a condition of supersaturation. However, because the seed crystals are already present, deposition of mineral salt from the supersaturated solution is minimised. The additional features of converting the precursor liquid into a supersaturated solution by cooling and/or reducing the pressure of the aqueous precursor liquid or by adding a complimentary ion into the precursor liquid (both of these are after addition of seed crystals) are likewise not disclosed or suggested by EP 916622. Thus, there would have been no motivation for the skilled person to modify EP 916622 to arrive at the presently claimed method because the method of EP 916622 works satisfactorily in the field for which it is described, name paper-making. It would, therefore, not have been obvious to one of ordinary skill based on EP 916622 that a substantial improvement in the reduction of deposition of mineral salts could be obtained in fields such as the oil industry by employing the presently claimed process.

Gatumel does cure the deficiencies of EP 916622 in regard to the presentity claimed invention. Gatumel is relied upon for an alleged disclosure relating to nucleation process for producing barium sulfate crystals using ultrasonic vibrations. Other than this, Gatumel is not relevant to the invention as claimed and does not give rise to a prima facie case of obviousness, either when taken alone or in combination with EP 916622.

In light of the above, it is clear that one of ordinary skill in the art would not have been motivated to arrive at the presently claimed invention based on the combined disclosures of the prior art relied on by the Examiner. Absent any such motivation, it is

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Reconsideration and withdrawal of the outstanding obviousness rejection of claims 8-16, 18, 25-29 and 32 are accordingly respectfully requested.

Claim 17 stands rejected under 35 U.S.C. §103(a) as allegedly unpatentable over EP '622 in view of Gatumel et al and further in view of Baumgard '966. This rejection is respectfully traversed.

Claim 17 has been amended to specify distribution of the composition into the aqueous precursor liquid being performed 2 to 4 times with the distribution being in series or parallel or a combination of both. The combined disclosures of EP '622 and Gatumel et al do not suggest claim 17 (which is dependent on claim 8) for the above discussed reasons. Baumgard does not cure the deficiencies of the two primary references relied upon to reject claim 17. The art relied upon does not suggest distribution of the composition into the aqueous precursor liquid being performed 2 to 4 times with the distribution being a series or parallel or a combination of both. Withdrawal of the outstanding rejection of claim 17 is accordingly respectfully requested.

II. ALLOWABLE SUBJECT MATTER

It is noted, with appreciation, that claim 30 is free of the prior art. Based on the amendments and arguments presented above, it is believed that all of the claims in this application are now in condition for allowance. Early notice to that effect is respectfully requested.

Favorable action on the application is awaited.

ACTON et al Appl. No. 10/014,795 July 21, 2004 PATENT MAINTENANCE DIVISION

Respectfully submitted,

2004 AUG 10 PH 2: 40

NIXON & VANDERHYE P.C.

Ву

Leonard C. Mitchard Reg. No. 29,009

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> Atty Dkt.: 3948-4 C#

Examiner: Kuhar, Anthony J.

Date: July 21, 2004

TC/A.U.: 1754

M#

In re Patent Application of

ACTON et al

Serial No. 10/014,795

Filed: December 14, 2001

Title:

REDUCTION IN MINERAL SALT DEPOSITION

Commissioner for Patents P.O. Box 1450 Alexandria, VA 22313-1450

Sir:

☒ Correspondence Address Indication Form Attached.

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		BRIEF is attached in triplicate in the pending appeal of the tified application (\$ 330.00)	\$	
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Adjustment date: 09/14/2004 EEKUBAY1 07/20/2004 JBALINAN 00000057 10014795 01 FC:1401

NIXON & VANDERHYE P.C.

By Atty: Leonard C. Mitchard, Reg. No. 29,009

Signature:

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IN THE UNITED STATES PATENT AND TRADEMARK OFFICE

In re Patent Application of

ACTON et al

Atty. Ref.: 3948-4; Confirmation No. 4355

Appl. No. 10/014,795

TC/A.U. 1754

Filed: December 14, 2001

Examiner: Kuhar, Anthony J.

For: REDUCTION IN MINERAL SALT DEPOSITION

July 21, 2004

Commissioner for Patents P.O. Box 1450 Alexandria, VA 22313-1450

Sir:

AMENDMENT

In response to the Official Action mailed January 21, 2004 (for which petition is hereby made for a three-month extension of time), please amend the above-identified application as follows:

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02_EC+1253

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Adjustment date: 09/14/2004 EEKUBAY1 07/20/2004 JBALINAN 00000057 10014795 02 FC:1253 -950.00 8P

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